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atomic percent oxygen. Consider a solid solution in the system Ti-O containing an atom fraction, $X_o = 0.12$.

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4.1 1st Law of

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The first law of

thermodynamics is

that energy is

conserved. This

means that the

increase in the

internal energy of the

system is equal to the

sum of the heat flow

and the work done on

the system and the

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heat flow into the
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dV and dS as a

function of T and P

that we derived in

class, fully derive the

rest of toolbox 3 (i.e.,

dU , dH , dF , and dG).

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– 74, equations 4.41 –

4.43. 1

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