

Student Exploration Ionic Bonds

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Although the chemical bonds within a water molecule are very stable, it's possible for a water molecule to give up a hydrogen and become OH^- , thus acting as a base, or accept another hydrogen and become H_3O^+ , thus acting as an acid. This adaptability allows water to combat drastic changes of pH due to acidic or basic substances in the ...

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This lecture course introduces students to various topics in inorganic chemistry. The topics covered include atomic structure, symmetry and group theory, introduction to ionic and covalent bonding models in coordination complexes, acid-base theories, aqueous chemistry, electrochemistry, and an introduction to bioinorganic chemistry.

Ionic Bonds Gizmo : ExploreLearning

If a question is asking about the high melting point of magnesium oxide, a student will not get much credit for just saying it has strong bonds or strong ionic bonds. A good answer might be: MgO has a giant lattice structure with strong electrostatic forces of attraction between oppositely charged ions. These require a lot of energy to break.

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Simulate ionic bonds between a variety of metals and nonmetals. Select a metal and a nonmetal atom, and transfer electrons from one to the other. Observe the effect of gaining and losing electrons on charge, and rearrange the atoms to represent the molecular structure. Additional metal and nonmetal atoms can be added to the screen, and the resulting chemical formula can be displayed.

Iowa Research Online

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Atoms, molecules and ions | CPD | RSC Education

Ionic compounds form when a positive metal ion bonds with a negative nonmetal ion. Of the p-block metals, several have fascinating properties. Gallium, in the 3rd row of column 3A, is a metal that ...

P-Block Elements on the Periodic Table: Properties ...

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By contrast, ionic compounds are made up of an indeterminate number of ions, in a fixed ratio.

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The particle of the ionic substance remains the ion. Using hands-on models can help students with these tricky concepts – eg TIMSTAR MO84200 for molecules and Molymod MKO-127-27 for ionic structures.

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2013 Killam Award for Excellence in Graduate Student Mentorship (UBC) 2013 Japan Society for the Promotion of Science (JSPS) Invitational Fellowship (NIMS, Japan) 2013 Strem Award for Pure or Applied Inorganic Chemistry (Canadian Society for Chemistry) 2012-14 E.W.R. Steacie Memorial Fellowship (NSERC)

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