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Name Formation of Ions

1. What are the names of the two elements shown? Date Class 22

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Use with Chapter 7,
Section 7.1 2. Are the
elements shown on the
left sides of the two
equations neutral?

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Section/Objectives

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**Standards Lab and
Demo Planning**

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WORKSHEET Isotopes

1. What do the following symbols represent?
b. no 2.

Which subatomic particles are found in an atom's nucleus?
3.

Which subatomic particle identifies an atom as that of a

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particular element? 4.
Explain why atoms are
neutral even though they
contain charged
particles. 30 5.

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Chapter 7, Section 7.1
and Predicting Ionic
Change 1. What
happens to a neutral
atom if it loses one
electron?

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Balanced ...

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Ions Section 7.1 1. What

are the names of the two

elements shown?

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8 Math Skills

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CHAPTER 3

Matter—Properties and Changes

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Chapters 21–25

Resources

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WORKSHEET Ionic

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Answer Key
1. How many valence electrons

does a neutral

magnesium (Mg) atom

have? 2. What is the

charge on a magnesium

ion? What does

magnesium have to do

to form such an ion, and

why does it tend to do

so? 3. How many

valence electrons does a

single neutral chlorine

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(CD atom have? 4.

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WORKSHEET Atomic

Orbitals 1. What is the
shape of an s orbital?

Class 16 Use with

Chapter 5, Section 5.2 2.

What is the relationship
between the size of an s

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Answers Key

orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital?

How many p orbitals are there in a sublevel? 4.

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WORKSHEET 31
ANSWERS PDF**

See page 14T for a key to the standards.

Differentiated

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activities should be
appropriate for students
with learning

difficulties. ... Mini Lab
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18-3 Connecting Math
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FILE Chapters 16–20

Resources, Chapter 18

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Master, p. 95

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8 Transparency
Worksheet 7Earth
Science: Geology, the
Environment, and the
Universe Teaching
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Class Date 1. How can
you determine the
number of protons in the
nucleus of an atom of

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any of the elements
listed in the table? 2.

Which element has 14
protons in the nuclei of
its atoms? 3.

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AND ANSWERS 12. c
13. salt 14. air 15.
aluminum

Understanding Main
Ideas (Part B) 1.

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- physical 2. chemical 3.
- chemical 4. physical 5.
- chemical 6. physical 7.
- chemical 8.
- homogeneous 9.
- heterogeneous 10.
- homogeneous 11.
- heterogeneous 12.
- homogeneous 13.
- heterogeneous 14.
- heterogeneous 15.
- heterogeneous 16.
- heterogeneous 17.
- homogeneous 18.

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homogeneous 19.
distillation 20 ...
Answer Key

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WORKSHEET The
Activity Series Use with
Chapter 9, Section 9.2 1.
For each of the
following pairs of
elements, underline the
one that would replace

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the other element in a
compound. a. calcium,
tin e. iron, copper b.
bromine, fluorine f.
iodine, chlorine c.
aluminum, potassium g.
silver, lead d. zinc,
sodium 2.

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