

## Ts Grewal Accountancy Cl 12 Solution Company File Type

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### Hybridization of BeCl<sub>2</sub> - Hybridization of Be and Cl in BeCl<sub>2</sub>

Which Of The Following Will Have The Maximum Dipole Moment? (A) CH<sub>3</sub>I. (B) CH<sub>3</sub>F. (C) CH<sub>3</sub>Br. (D) CH<sub>3</sub>Cl. Get the answer to this question and access other important questions, only at BYJU'S.

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What is the Hybridization of Beryllium Dichloride? To know about the hybridization of BeCl<sub>2</sub> (Beryllium Dichloride) we have to take a closer look at the central atom which is Be. Its electronic configuration is 1s<sup>2</sup>, 2s<sup>2</sup>, where two electrons are present in the valence shell. During the formation of BeCl<sub>2</sub>, beryllium atom bonds with two chlorine atoms via single covalent bonds.

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