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Hybridization of BeCl₂ - Hybridization of Be and Cl in BeCl₂

Which Of The Following Will Have The Maximum Dipole Moment? (A) CH₃I. (B) CH₃F. (C) CH₃Br. (D) CH₃Cl. Get the answer to this question and access other important questions, only at BYJU'S.

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What is the Hybridization of Beryllium Dichloride? To know about the hybridization of BeCl₂ (Beryllium Dichloride) we have to take a closer look at the central atom which is Be. Its electronic configuration is 1s², 2s², where two electrons are present in the valence shell. During the formation of BeCl₂, beryllium atom bonds with two chlorine atoms via single covalent bonds.

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