

### **Types Of Chemical Reactions Pogil Answer Key**

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#### **35 Types Of Chemical Reactions Worksheet Pogil ...**

Four common types are composition, decomposition, single replacement

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and ionic reactions. Specific reactions corresponding to these general types are associated with health issues, environmental problems, and manufacturing processes. In order to use chemical reactions or evaluate their effects, you need to be able to identify the type of reaction.

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### **POGIL® Activities for High School Chemistry**

Reactions – Period 5 Edition Chemical reactions can be classified into different categories. Four common types are synthesis, decomposition, single replacement, and double replacement. Specific reactions corresponding to these general types are associated with health issues, environmental problems, and manufacturing processes. In order to use chemical reactions or

### **Classifying Types of Chemical Reactions (POGIL ...**

a. Write the chemical reaction in reverse. b. Label the reaction

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written in part a with one of the reaction types in Model 1. 13.  
Identify each of the reactions below as synthesis (S), decomposition (D), single replacement (SR) or double replacement (DR).  
$$\text{ISO}(s) + \text{H}_2\text{O}(l) \rightarrow 2\text{KOH}(aq)$$
$$2\text{NaCl}(aq) + \text{MgCO}_3(s) \rightarrow 2\text{MgCl}_2(aq) + \text{Na}_2\text{CO}_3(aq)$$
$$4\text{Al}(s) + 3\text{O}_2(g)$$

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Types of Chemical Reactions POGIL - Types of Chemical... Types of Chemical Reactions Model 1: Synthesis and Decomposition Reactions  
Consider the followings synthesis and decomposition reactions:  
Examples of Synthesis Reactions:  $\text{Cu} + \text{I}_2 \rightarrow \text{CuI}_2$ ,  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$   
 $4\text{Na} + \text{O}_2 \rightarrow 2\text{Na}_2\text{O}$  or  $\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4\text{Cl}$  ;  $\text{NI} + 4\text{CO}_3 \rightarrow 2\text{Na} + \text{S} + 2\text{O}_2 \rightarrow \text{Na}_2\text{SO}_4$   
Examples...

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There are 5 general types of chemical reactions. Those types are synthesis (or direct combination), decomposition, single replacement, double replacement (or metathesis), and combustion.

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### **POGIL Activity: Types of Chemical Reactions Summary**

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### **POGIL 03 - Chemical Reactions 3 - Types of Reactions**

Similarly, chemists classify chemical equations according to their patterns to help predict products of unknown but similar chemical reactions. 1. Match each description below to one of the reactions sets (A, B, C, or D) from Model 1.

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Flinn Scientific and The POGIL® Project have collaborated to publish this series of student-centered learning activities for high school chemistry. Create an interactive learning environment with 36 specially designed guided-inquiry learning activities in 13 major

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topic areas. 290 pages, 8½" x 11".

### **Types Of Chemical Reactions Pogil**

The chemical equations in Model 1 contain the phase notations G), 0), G), and (aq). Match each symbol with its meaning. dissolved (\*r) in water liquid (-e) solid (s) gas (g) Based on the examples provided, which set(s) of reactions in Model 1 typically involve ions in solution (A, B, C or D)? 1\ - -1-\ \- -Lj

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In this activity we learned how to identify types of chemical reactions by looking at written-out chemical equations. Just as living things can be classified into groups such as mammals, plants, and fish chemical reactions can be classified as well. In this activity there were four types of reactions.

### **21 Types of Chemical Reactions-S**

In order to use chemical reactions or evaluate their effects, you need to be able to identify the type of reaction. Success Criteria • Identify and differentiate between four types of chemical reactions: synthesis, decomposition, single replacement and double replacement.

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materials. An exploratory paper isn't unusual in businesses when they  
will need to receive all of the feasible perspectives and're trying to  
have a remedy and data available.

### **POGIL Chemistry Teachers Edition**

4 POGIL™ Activities for High School Chemistry 15. A student writes the  
following incorrect chemical equation for a single replacement

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reaction between lithium bromide and fluorine.  $2\text{LiBr}(\text{aq}) + \text{F}_2(\text{g}) \rightarrow 2\text{Li}(\text{s}) + 2\text{FBr}(\text{g})$  a. In a single replacement reaction, part of an ionic compound is removed and replaced by a

### **Types of Chemical Reactions POGIL – Types of Chemical ...**

chemical reactions or evaluate their effects, you need to be able to identify the type of reaction. Success Criteria • Identify and differentiate between four types of chemical reactions: synthesis, decomposition, single replacement and double replacement.

Prerequisites • Reaction equation nomenclature • Balancing equations  
New Concepts • Syntheses • Decomposition • Single replacement • Double replacement © POGIL – 2005 1/6

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