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**Naming & Formulas
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A gas is a state of

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matter with no defined shape or volume.

Gases have their own unique behavior depending on a variety of variables, such as temperature, pressure, and volume. While each gas is different, all gases act in a similar matter.

This study guide highlights the concepts and laws

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the following
decreases as the
strength 7. On a
relative

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of the composition,
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Unit Analysis Goals

To describe a
procedure for making
unit conversions
called unit analysis.

To describe metric-
metric unit

conversions. To
describe English-
metric unit

conversions. Many

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chemical calculations include the conversion from a value expressed in one unit to the ...

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Chemical Reactions

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g e The law of

conservation of mass

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states that the mass of the reactants will always equal the mass of the products. In result the number of atoms of one element on the reactants side should be identical to the number of atoms of the same element on the products side.

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Energy is the ability to do work or produce heat 2. Energy cannot be created or destroyed it can only be transformed . 3. The energy in the universe is constant. 4. A state function changes independent of the pathway. 5. Temperature is a measure of the random motion of the

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components of a
substance. 6. Heat is
measured in ...

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Chemistry II Honors**

Unit 13 Study ...

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Unit 8- Chemistry

General

Announcements:

Answer to the In-

Class Review Sheet

... Chemistry Unit

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Exam: Thursday, April
2nd Chemistry Part 1

Study Guide Topics: -

States of matter and
examples of each

state - Law of

Conservation of

Matter - Chemistry

Vocabulary: volume,

mass, density, etc. -

Calculating Density

(formula $D=m/v$)

Chemistry | Science

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Completing this unit should take you approximately 8 hours. Unit 2: The Atom The atom, along with its protons, neutrons, and electrons, is the basic unit of matter and serves as our starting point for the study of chemistry.

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