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1900s scientists discovered that various isotopes will undergo nuclear decay. During this process the unstable nucleus of an atom gives off radiation. When scientists studied this radiation they discovered three types of particles: alpha, beta, and gamma.

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chemistry dealing with radioactivity, nuclear processes, such as nuclear transmutation, and nuclear properties. It is the chemistry of radioactive elements such as the actinides, radium and radon together with the chemistry associated with equipment (such as nuclear reactors) which are designed to perform nuclear

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thorium-231. 2. Lanthanum -144 becomes cerium-144 when it undergoes a beta decay. 3. Neptunium-233 is formed when americium-237 undergoes a nuclear decay process. 4.

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Changes of nuclei that result in changes in  
their atomic numbers, mass numbers, or

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energy states are nuclear reactions . To describe a nuclear reaction, we use an equation that identifies the nuclides involved in the reaction, their mass numbers and atomic numbers, and the other particles involved in the reaction.

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